## SPEAKER

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## Problems of room temperature superconductivity and metallic hydrogen

**ABSTRACT** • Metallic hydrogen and room-temperature superconductivity are one of the most challenging and very long standing problems in solid-state physics. In both, there is a significant progress over the recent years. Nearly room temperature conventional superconductivity was discovered in hydrides<sup>1</sup>: recently the critical temperature Tc ~250 K at high pressure of ~150 GPa was found in superhydride LaH10<sup>213</sup> following the predictions<sup>415</sup>.

We will discuss prospects for further increase of Tc to room temperature, in particular in yttrium hydride YH10 with predicted Tc ~300 K at 400 GPa and Li2MgH16 with Tc of ~473K at 250 GPa. We will consider various directions to explore high temperature conventional superconductivity at low and ambient pressures. Metallic hydrogen was predicted by Wigner and Huntington<sup>6</sup> in 1935, and N. Ashcroft in 1968 suggested that it should be a high temperature superconductor<sup>7</sup>, room temperature superconductivity follows from the recent calcultions<sup>7+8</sup>. However achieving and measuring of metallic hydrogen is very challanging task, in particular, very high pressures are required<sup>9</sup>: only at ~370-500 GPa solid molecular hydrogen would dissociate and form atomic solid at pressures.

In another scenario, the metallization first occurs in the 250-500 GPa pressure range in molecular hydrogen through overlapping of electronic bands<sup>10</sup>. The calculations are not accurate enough to predict which option is realized. Our experiments<sup>11</sup> indicate the metallization in molecular hydrogen through closing of energy gap. We observed that at a pressure of ~360 GPa and temperatures <200 K (phase III) the hydrogen starts to conduct, and that temperature dependence of the electrical conductivity is typical of a semimetal.

Raman spectra, measured up to 480 GPa, indicate that hydrogen remains a molecular solid at pressures up to 440 GPa, while at higher pressures the Raman signal vanishes, likely indicating further transformation to a good molecular metal or to an atomic state.

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<sup>10</sup> Johnson, K. A. & Ashcroft, N. W. Structure and bandgap closure in dense hydrogen. Nature 403, 632-635 (2000).

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